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Determination of a novel selective inhibitor of type 1 5α -reductase in human plasma by liquid chromatography with atmospheric pressure chemical ionization tandem mass spectrometry

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Abstract

A sensitive and specific assay of human plasma for the determination of $(5\alpha,7\beta,16\beta)$ -16[(4-chlorophenyl)oxy]-4,7dimethyl-4-aza-andronstan-3-one (**I**), a selective inhibitor of human type 1 5a-reductase, has been developed. The method is based on high-performance liquid chromatography (HPLC) with tandem mass spectrometric (MS–MS) detection. The analyte (**I**) and internal standard, Proscar (**II**), were isolated from the basified biological matrix using a liquid–liquid extraction with methyl-*tert*.-butyl ether (MTBE). The organic extract was evaporated to dryness, the residue was reconstituted in mobile phase and injected into the HPLC system. The MS–MS detection was performed on a PE Sciex API III Plus tandem mass spectrometer using a heated nebulizer interface. Multiple reaction monitoring using the precursor→product ion combinations of *m*/*z* 430→114 and 373→305 was used to quantify **I** and internal standard (**II**), respectively. The assay was validated in the concentration range of 0.5 to 500 ng/ml in human plasma. The precision of the assay, expressed as coefficient of variation (C.V.), was less than 7% over the entire concentration range, with adequate assay specificity and accuracy. The HPLC–MS–MS method provided sufficient sensitivity to completely map the 24 h pharmacokinetic time-course following a single 0.5 mg dose of **I**. \circ 1998 Elsevier Science B.V. All rights reserved.

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oxy]-4,7-dimethyl-4-aza-andronstan-3-one (L-751 Finasteride has undergone successful clinical evaluachemical structure similar to finasteride (Proscar), a benign prostatic hyperplasia (BPH). potent 5 α -reductase (5 α R) inhibitor [1]. Inhibition of The two 5 α R isozymes have been characterized 5aR results in an antiandrogen effect by decreasing and both are found in human skin in a ratio that target-organ dihydrotestosterone (DHT) levels. Two varies depending upon location. Beard and scrotal isozymes of $5\alpha R$ are known to exist, type 1 ($5\alpha R1$) skin contain mainly $5\alpha R2$ whereas chest skin con-

1. Introduction and type 2 ($5\alpha R2$). In contrast to finasteride, which is a potent inhibitor of $5\alpha R2$, compound **I** is a potent Compound **I** $(5\alpha,7\beta,16\beta)-16[(4-chlorophenyl)-$ and specific inhibitor of the human $5\alpha R1$ [2,3]. 788, Fig. 1) belongs to a class of 4-azasteroids with a tion [4], and was approved for use in the treatment of

tains both isozymes. Scalp, face, arm, breast, ab- *Corresponding author. \blacksquare domen, back and legs contain mainly $5\alpha R1$. The

localization of the $5\alpha R1$ in the sebaceous gland of demonstrated. skin [5] and the implication that DHT is one of the prerequisites for the onset of acne suggests that inhibition of this isozyme may be useful in the **2. Experimental** management of this disorder. Therefore, **I**, an inhibitor of 5aR1, may be useful in the prevention or 2.1. *Materials* treatment of acne.

studies of **I**, it was necessary to develop a sensitive Merck Research Labs. (Rahway, NJ, USA). All and specific method for the determination of **I** in solvents and reagents were of HPLC or analytical human plasma with the limit of quantification (LOQ) grade and were purchased from Fisher Scientific of less than 1 ng/ml. The LOQ target of 1 ng/ml (Fair Lawn, NJ, USA). The drug-free, human was established based upon historical data of finas- heparinized plasma originated from Biological teride and the projected oral doses of **I**. Specialties (Lansdale, PA, USA). Air (hydrocarbon-

performance liquid chromatography (HPLC) assay purchased from West Point Supply (West Point, PA, with a conventional ultraviolet (UV) absorbance USA). detection was evaluated. The presence of chlorophenoxy group in **I** created the possibility of an 2.2. *Instrumentation* improved sensitivity of detection in the near UV region due to the enhanced absorption of **I** in A PE Sciex (Thornhill, Canada) API III Plus

The UV absorbance spectra of **I** in methanol indicated the presence of two absorption bands with the maxima at 204 nm and 230 nm and the molar absorptivities (ϵ) of 18 600 and 14 700 M^{-1} cm⁻¹. respectively. In principle, the development of an assay based on HPLC with UV absorbance detection at 230 nm with the LOO of \sim 1 ng/ml was feasible but would require a very highly selective and efficient off-line and on-line sample clean-up using column switching to monitor **I** at sub-nanogram concentrations without endogenous interferences from plasma matrix. Instead, we have decided to use HPLC–tandem mass spectrometry (MS–MS) to achieve the LOQ of 0.5 ng/ml using relatively simple sample preparation procedure and short analysis time. The HPLC–MS–MS method was utilized earlier for determination of structurally similar azasteroids in both clinical and preclinical studies [7– 10].

The subject of this paper was the development of a highly sensitive HPLC–MS–MS assay for the determination of **I** in human plasma with the LOQ of 0.5 ng/ml. The need for careful assessment of the specificity of MS–MS based assays [11,12] was also Fig. 1. Chemical structures of **I** and internal standard **II**. emphasized and the absence of a matrix effect and assay specificity in the presence of metabolites was

In order to support initial clinical pharmacokinetic Compounds **I** and **II** were synthesized at the Initially, the feasibility of development of a high- free), nitrogen (99.999%) and argon (99.999%) were

comparison with other azasteroids studied earlier [6]. tandem mass spectrometer equipped with heated

nebulizer interface, a Waters Associates (Waters- curves using linear least square regression of the Millipore, Milford, MA, USA) WISP 715 autoin- plasma concentrations and the measured peak area jector, and Perkin-Elmer biocompatible binary pump ratios. Data collection, peak integration and calcula- (Model 250) were used for all HPLC–MS–MS tions were performed using MacQuan PE-Sciex analyses. The data were processed using MacQuan software. software (PE Sciex) on a Macintosh Quadra 900 microcomputer. 2.5. *Standard solutions*

Scientific BDS Hypersil C_{18} , 50×4.6 mm I.D., 5 μ m the concentrations of 5, 10, 50, 100, 500, 1000 and analytical column heated to 60°C and coupled with a 5000 ng/ml for plasma assay. The internal standard analytical column heated to 60° C and coupled with a 5- μ m in-line filter. The aqueous portion of the **II** was also prepared as a stock solution (1 mg/ml) mobile phase was prepared by dissolving 0.77 g of in methanol by dissolving 10 mg of solid **II** in 10 ml ammonium acetate in 1000 ml of water and the of methanol. A working standard of 100 ng/ml was addition of 820 μ l formic acid. The mobile phase prepared by dilution of stock standard with methawas a mixture of 80% acetonitrile and 20% water and mol, and was used for all analyses. All standards containing 0.1% formic acid and 10 mM ammonium were prepared once a month and stored at 5° C. acetate, and was delivered at a flow-rate of 1.0 A series of quality control (QC) samples at 1 and ml/min. The retention times for **I** and **II** were 2.8 400 ng/ml for the plasma assay were prepared. and 0.8 min. Aliquots (1.25 ml) of these solutions were placed in

was interfaced via a Sciex heated nebulizer probe to the HPLC system, and gas phase chemical ionization 2.6. *Sample preparation* was effected by a corona discharge needle $(+4 \mu A)$ using positive ion atmospheric pressure chemical A 1-ml aliquot of plasma was pipetted into a ionization (APCI). The heated nebulizer probe tem- 15-ml centrifuge tube and 100 μ l of the working perature was maintained at 500° C. The nebulizing standard of **II** (equivalent to 10 ng/ml of I.S.) gas (air) pressure and auxiliary flow were set at 550 followed by the addition of 1 ml of 0.2 *M* carbonate kPa and 2.0 l/min, respectively. Curtain gas flow buffer (pH 9.8). After addition of 7 ml of methyl- (nitrogen) was 0.9 l/min, and the sampling orifice *tert*.-butyl ether and capping tubes with PTFE-lined potential was set at $+65$ V. The dwell time was 400 caps, the mixture was rotated and mixed for 15 min. ms, and the temperature of the interface heater was The tubes were then centrifuged and the organic set at 60°C. The mass spectrometer was programmed layer was transferred to a clean centrifuge tube. The to admit the protonated molecules $[M+H]^+$ at m/z organic extract was evaporated to dryness under a 430 (**I**) and m/z 373 (**II**), via the first quadrupole stream of nitrogen at 50°C, the residue was reconstifilter (O1), with collision-induced fragmentation at tuted in 300 μ l of the mobile phase and a 150- μ l filter (Q1), with collision-induced fragmentation at tuted in 300 μ l of the mobile phase and a 150- μ l Q2 (collision gas argon, 270·10¹³ atoms cm⁻²), and aliquot was injected onto the HPLC–MS–MS sysmonitoring the product ions via Q3 at m/z 114 and tem. 305 for **I** and **II**, respectively. The electron multiplier setting was 24.7 kV. Peak area ratios obtained from 2.7. *Precision*, *accuracy and recovery* multiple reaction monitoring of the analyte and internal standard, $(m/z \ 430 \rightarrow 114)/(m/z \ 373 \rightarrow 305)$, The precision of the method was determined by were utilized for the construction of calibration the replicate analyses $(n=5)$ of human plasma con-

2.3. *Chromatographic conditions* A stock solution of **I** (1 mg/ml) was prepared in methanol. This solution was further diluted with HPLC separation was performed using a Keystone methanol to give a series of working standards with

2-ml plastic tubes, stored at -20° C, and analyzed 2.4. *HPLC*–*MS*–*MS conditions* daily with clinical samples. The calculated concentrations of the QC samples were compared on a A PE Sciex triple quadrupole mass spectrometer day-to-day basis to assess inter-day assay variability.

ing calibration curves. The linearity of each standard selected post-dose samples were assayed under curve was confirmed by plotting the peak area ratio $HPLC$ gradient elution conditions, thereby greatly of the drug to I.S. versus drug concentration. The increasing the retention factor (k) for both analytes. of the drug to I.S. versus drug concentration. The increasing the retention factor (*k*) for both analytes. from the equation *y*=*mx*+*b*, as determined by dient analysis, when monitored at the pre-
weighted $(1/\gamma^2)$ linear regression of the standard cursor→product ion combinations used for the deline. The standard curve was constructed daily and termination of **I** and **II**, indicating the metabolites of these standard samples were assayed with quality **I** have not interfered with the quantification of **I** and control and unknown samples. The accuracy of the **II**. method was expressed by [(mean observed concentration)/(spiked concentration) $]\times 100$. The recovery was determined by comparing the peak area **3. Results and discussion** of **I** extracted from biological fluids to that of standards injected directly. 3.1. *Assay validation*

liability is critical when homologues rather then 114. For the internal standard **II**, the fragmentation stable isotope-labeled parent compound is selected was more favorable, and the product mass spectrum for use as internal standard [11,12]. The undetected of the protonated molecule of **II** at m/z 373 exbut co-eluting endogenous impurities may affect the hibited only two major fragments at *m*/*z* 317 and ionization efficiencies of the analytes. Therefore, the 305 (Fig. 2) that is generally much more favorable ionization efficiency of both analytes, as measured for sensitive MS–MS detection. by the individual peak areas of **I** and **II** in different Initially, four different precursor→product ion

samples, after dosing. In addition, in order to assess multiple reaction monitoring mode, a highly sensi-

taining **I** at all concentrations utilized for construct- assay specificity in the presence of metabolites, No additional peaks were observed during the gra-I have not interfered with the quantification of I and

2.8. *Assessment of matrix effect* The positive product mass spectrum of the proton-
ated $[M+H]$ ⁺ molecule of **I** at m/z 430 indicated the The assessment of matrix effect and assay re-
presence of several ions at m/z 271, 253, 229 and

plasma matrix, was evaluated. Both **I** and **II** were combinations (*m*/*z* 430→271, 430→253, 430→229, spiked into five different sources of plasma and their 430→114) were evaluated for sensitive and selective peak areas were determined. By comparing peak determination of **I**. The intensity of response and the areas of the same analyte in different lots of plasma, presence of endogenous plasma impurities in differpresence of endogenous plasma impurities in differthe recovery and the differences in ionization ef- ent channels and at the retention time of **I** were ficiency associated with a given plasma lot was monitored in plasma samples originating from ten assessed. different subjects. Small peaks originating from endogenous plasma components were observed in 2.9. *Assessment of assay specificity* channels *m*/*z* 430→271, 430→253, 430→229 in at least one source of plasma, but such interference was Assay specificity for the clinical samples was not present in the channel m/z 430 \rightarrow 114. It was later assessed by running blank control and patients' pre- confirmed that the method was selective at *m*/*z* dose biological fluid samples. No endogenous inter- 430→114 and interference peaks in pre-dose plasma ferences were observed. In addition, internal standard samples from all subjects participating in clinical peak areas in plasma samples of twelve subjects studies were not observed. Therefore, concentrations participating in a clinical study were compared for of **I** were determined at m/z 430 \rightarrow 114 over other reproducibility at all timepoints after dosing with **I**. MS–MS channels available. For internal standard The coefficient of variation (C.V.) of the internal **II**, the most intense product ion at m/z 305 was standard peak areas was 6.3%, further indicating selected for quantification. By monitoring the consistent recovery and ionization efficiency of the precursor→product ion combinations at m/z internal standard between various subject timepoint $430\rightarrow114$ for **I** and m/z 373 \rightarrow 305 for **II** in the

Fig. 2. Positive product ion mass spectra of the protonated grams are shown in Fig. 3.
molecules of **I** at m/z 430 (A) and internal standard **II** at m/z 373 The potential ion suppress

was developed. Were practically the same (C.V. of peak areas was

on a simple liquid–liquid extraction from basified difference in ionization efficiency and consistent plasma, evaporation of the extract to dryness, recon- recovery of analytes **I** and **II** from five different stitution of the residue in mobile phase and injection plasma sources. In addition, any small changes in into the HPLC system. The sample was basified to peak area of **I** in different lots of plasma were reduce the amount of compounds extracted from the compensated by small similar changes in the peak matrix, thereby reducing the endogenous material areas of the internal standard **II**. The C.V. of peak injected onto the analytical column and increasing area ratios of I/II was much smaller than the C.V.s of column lifetime and possibly reducing matrix effects. peak areas of **I** and **II**, confirming the desired

reduce the retention time of analytes and to improve standard on the precision and reliability of the peak shape. Similar decrease in the retention time quantification of **I**. In addition, the reproducibility can be achieved by increasing acetonitrile content in and accuracy of the determination of internal stanthe mobile phase from 80 to 95%, but under these dard was highly adequate as demonstrated in Table

conditions the internal standard **II** was practically unretained and eluted at 0.5 min. In addition, the efficiency of ionization of **I** and **II** was lower when a mobile phase with high organic content (95% acetonitrile) was utilized in comparison with the efficiency of ionization when a mobile phase with lower organic content (80% acetonitrile) was used. Therefore, in order to achieve the desired assay sensitivity, column heating and a mobile phase with a lower organic content rather than a mobile phase with higher organic content was employed. Heating of the analytical column had no adverse effect on its performance, as demonstrated by the analysis of more than 1200 samples from clinical studies using a single analytical column.

Following the procedure described in Section 2, the assay was validated in human plasma in the concentration range of 0.5 to 500 ng/ml. The difference between the nominal standard concentration and the back-calculated concentration from the weighted linear regression line was less than 10% for each point on the standard curve indicating that the linear regression analysis applied $(1/y^2)$ provided an adequate fit of the data. The correlation coefficient was always greater than 0.99. For ten daily runs, the mean slope and correlation coefficients of the calibration lines were 0.00777 ± 0.00059 and 0.996 ± 0.002 , respectively. Representative chromato-

molecules of 1 at m/z 430 (A) and internal standard **H** at m/z 3/3
(B). The potential ion suppression effect and ionization efficiencies for **I** and **H** in plasma samples originating from different individuals were closely extive assay for **I** in plasma with the LOQ of 0.5 ng/ml amined. As seen in Table 1, peak areas of **I** and **II** The isolation of **I** and **II** from plasma was based less than 10%), strongly indicating little or no The analytical column was heated to 60° C to compensating effect of the presence of internal

Fig. 3. Representative HPLC–MS–MS chromatograms of plasma (1 ml) extracts obtained by multiple reaction monitoring at *m*/*z* 430→114 (channel 'a') for **I** and m/z 373→305 (channel 'b') for internal standard (II); (A, A') Blank control plasma monitored at channels 'a' and 'b', respectively; (B, B') control plasma spiked with 1 ng of **I** and 10 ng of **II** monitored at channels 'a' and 'b', respectively; (C, C') pre-dose plasma sample of a subject spiked with 10 ng **II** and monitored at channels 'a' and 'b', respectively; (D, D') plasma sample of a subject 24 h after receiving a 0.5 mg oral dose of **I** and spiked with 10 ng of **II** monitored at channels 'a' and 'b'; concentration equivalent to 0.62 ng/ml. The numbers in the upper right hand corner correspond to peak heights expressed in arbitrary units.

1. In spite of the fact that internal standard was only were practically the same (Table 2). Also, the peak modestly retained on the column (retention factor, areas of **II** in post-dose plasma samples were very $k \sim 1$), an in-depth evaluation of a potential matrix similar to each other and to the peak areas in control effect using control plasma samples from a number plasma confirming that the potential interferences of individuals indicated that the peak areas of **II** from metabolites present in post-dose biological

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Intra-day precision of peak areas of **I** and **II** spiked into five different sources of plasma

^a I spiked into five different sources of plasma, in arbitrary units.

^b **II** spiked into five different sources of plasma, in arbitrary units.

^a Mean concentrations calculated from the weighted linear least-squares regression curve constructed using all five replicate values at each concentration.

 b Expressed as coefficient of variation (C.V., %).

Expressed as [(mean observed concentrations/nominal concentration) \times 100] (*n*=5).

^d Recovery was determined by comparing the peak area of **I** extracted from biological fluids to that of standards injected directly.

these data illustrate the adequacy of choosing **II** as indicating adequate stability of **I** in the biological an internal standard in spite of its relatively limited matrix frozen at -20° C. retention $(k¹)$ on the column. Stability of **I** during multiple freeze–thaw cycles

less than 9% at all concentrations within the standard samples. Following three freeze–thaw cycles the curve range (Table 2). The mean recovery of **I** from mean values $(n=2)$ were 0.98 ± 0.16 and 419.8 ± 23.9 plasma was 98% and was practically the same at all for low and high QC samples, respectively. concentrations within the standard curve range (Table 2). 3.2. *Analyses of samples from clinical studies*

The limit of quantification (LOQ) of the assay was 0.5 ng/ml. The LOQ was defined here as the lowest The performance and ruggedness of the HPLC– concentration on the standard curve for which preci- MS–MS assay was tested by analyzing more than sion of the determination, expressed as C.V., was less 1200 plasma samples from a clinical study with **I**. As than 10%, with assay accuracy within $\pm 10\%$ of the an example, representative concentrations of **I** in nominal concentrations of **I**. plasma after oral administration of a 2 mg dose of **I**

a clinical trial are presented in Table 3. These data In conclusion, an assay for **I** in human plasma show good inter-assay accuracy and precision. In with the LOQ of 0.5 ng/ml was developed and its addition, both low and high QC samples were performance and ruggedness was tested during multianalyzed after three months of storage at -20°C . sample analyses of samples from clinical studies.

fluids were not affecting the peak area of **II**. All of Mean values $(n=5)$ were 1.04 \pm 0.13 and 404.2 \pm 8.23

The intra-day precision, expressed as the C.V. was was also evaluated by analyzing low and high QC

The summary of the analyses of QC standards for to selected human subjects are shown in Fig. 4.

Table 3

Inter-day accuracy and precision data for the assay of quality control samples spiked with **I**

Initial ^ª assay concentration (ng/ml)	No. of determinations	Mean calculated concentrations (ng/ml)	CN. (%)
1.00	16°	1.04	5.3
405.8	16 ^b	441.18	4.9

 $n^{a} n = 5.$

^b Over a period of eight days.

Assay selectivity and the absence of matrix effects, togr. 566 (1991) 127–132.
were demonstrated as required when an analog [7] M.L. Constanzer, C.M. Chavez, B.K. Matuszewski, J. Chrowere demonstrated, as required when an analog [7] M.L. Constanzer, C.M. Chavez, I.
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